

## 2-(Thiophen-2-yl)ethyl 4-methylbenzenesulfonate

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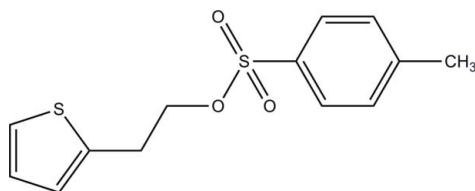
Received 31 May 2011; accepted 18 June 2011

Key indicators: single-crystal X-ray study;  $T = 113$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.035;  $wR$  factor = 0.088; data-to-parameter ratio = 19.3.

In the title molecule,  $\text{C}_{13}\text{H}_{14}\text{O}_3\text{S}_2$ , the thiophene and benzene rings form a dihedral angle of  $13.86$  ( $13$ )°. In the crystal, weak intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds link the molecules into layers parallel to the  $ab$  plane.

## Related literature

The title compound is an intermediate in the synthesis of the antiplatelet agent clopidogrel (systematic name (+)-(*S*)-methyl 2-(2-chlorophenyl)-2-(6,7-dihydrothieno[3,2-*c*]pyridin-5(4*H*)-yl)acetate). For background to the bioactivity and applications of clopidogrel, see: Raju *et al.* (2008). For the synthesis of the title compound, see: Sajja *et al.* (2007).



## Experimental

## Crystal data

$\text{C}_{13}\text{H}_{14}\text{O}_3\text{S}_2$   
 $M_r = 282.36$   
 Monoclinic,  $P2_1$   
 $a = 8.6130$  (9) Å

$b = 5.9961$  (4) Å  
 $c = 13.1284$  (12) Å  
 $\beta = 97.935$  (19)°  
 $V = 671.52$  (10) Å<sup>3</sup>

$Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.39$  mm<sup>-1</sup>

$T = 113$  K  
 $0.20 \times 0.18 \times 0.10$  mm

## Data collection

Rigaku Saturn CCD area-detector diffractometer  
 Absorption correction: multi-scan (*CrystalClear*; Rigaku/MS, 2005)  
 $T_{\min} = 0.926$ ,  $T_{\max} = 0.962$

6939 measured reflections  
 3167 independent reflections  
 2365 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.048$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$   
 $wR(F^2) = 0.088$   
 $S = 0.83$   
 3167 reflections  
 164 parameters  
 1 restraint

H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.39$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.38$  e Å<sup>-3</sup>  
 Absolute structure: Flack (1983),  
 1405 Friedel pairs  
 Flack parameter: 0.00 (8)

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$                          | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|-------|-------------|-------------|---------------|
| $\text{C}2-H2\cdots\text{O}2^i$        | 0.95  | 2.58        | 3.523 (3)   | 174           |
| $\text{C}6-H6B\cdots\text{O}2^{ii}$    | 0.99  | 2.41        | 3.161 (3)   | 132           |
| $\text{C}13-H13A\cdots\text{O}3^{iii}$ | 0.98  | 2.58        | 3.496 (3)   | 155           |

Symmetry codes: (i)  $-x + 1, y + \frac{1}{2}, -z + 1$ ; (ii)  $x, y + 1, z$ ; (iii)  $x - 1, y, z$ .

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *CrystalStructure* (Rigaku/MS, 2005).

The authors thank Mr Hai-Bin Song of Nankai University for the X-ray crystal structure determination and helpful suggestions.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5113).

## References

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**supplementary materials**

*Acta Cryst.* (2011). E67, o1787 [ doi:10.1107/S1600536811023907 ]

## 2-(Thiophen-2-yl)ethyl 4-methylbenzenesulfonate

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### Comment

Clopidogrel, a thienopyridine class inhibitor of P2Y<sub>12</sub> ADP platelet receptor, has been found to be particularly useful in the treatment of coronary artery disease, peripheral vascular disease and cerebrovascular disease (Raju *et al.*, 2008). Herewith we present the crystal structure of the title compound (I) used as an intermediate in the synthesis of clopidogrel (Sajja *et al.*, 2007).

In (I) (Fig. 1), the dihedral angle formed between the benzene ring plane (r.m.s. deviation 0.0029 Å) and the thiophene ring plane (r.m.s. deviation 0.0025 Å) is 13.86 (13)°. The packing of the crystal is consolidated by the weak C—H···O interactions (Table 1).

### Experimental

5 g of 4-Methylbenzene-1-sulfonyl chloride and 30 ml of toluene were charged into a clean and dry reactor followed by cooling to about 5 °C. 3.4 g 2-(Thiophen-2-yl)ethanol was added at about 5 °C over about 20 minutes, followed by addition of 4.5 g of triethylamine over about 6 h. The reaction mixture temperature was raised to about 30 °C, followed by stirring for about 9 h. The reaction mass was filtered through a Nutsche filter and washed with 50 ml of toluene, and then, the reaction filtrate was transferred into another reactor followed by washing with 80 ml of water. Organic and aqueous layers were separated and the organic layer was distilled completely at about below 70 °C to afford 6.3 g of off-white solid as crude product. The solid was dissolved in methanol 30 ml at 20 °C, then colourless crystals were generated slowly.

### Refinement

All H atoms were positioned geometrically (C—H 0.95 - 0.99 Å) and refined using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2$  or  $1.5U_{\text{eq}}(\text{C})$ .

### Figures

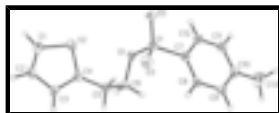


Fig. 1. The molecular structure of (I), with the atom-numbering scheme and 50% probability displacement ellipsoids.

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### Crystal data

C<sub>13</sub>H<sub>14</sub>O<sub>3</sub>S<sub>2</sub>

$M_r = 282.36$

$F(000) = 296$

$D_x = 1.396 \text{ Mg m}^{-3}$

# supplementary materials

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Monoclinic,  $P2_1$   
Hall symbol: P 2yb  
 $a = 8.6130$  (9) Å  
 $b = 5.9961$  (4) Å  
 $c = 13.1284$  (12) Å  
 $\beta = 97.935$  (19)°  
 $V = 671.52$  (10) Å<sup>3</sup>  
 $Z = 2$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å  
Cell parameters from 2573 reflections  
 $\theta = 2.4$ – $27.9$ °  
 $\mu = 0.39$  mm<sup>-1</sup>  
 $T = 113$  K  
Prism, colourless  
 $0.20 \times 0.18 \times 0.10$  mm

## Data collection

Rigaku Saturn CCD area-detector diffractometer  
Radiation source: rotating anode multilayer  
Detector resolution: 14.63 pixels mm<sup>-1</sup>  
 $\omega$  and  $\phi$  scans  
Absorption correction: multi-scan (*CrystalClear*; Rigaku/MSC, 2005)  
 $T_{\min} = 0.926$ ,  $T_{\max} = 0.962$   
6939 measured reflections

3167 independent reflections  
2365 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.048$   
 $\theta_{\max} = 27.9$ °,  $\theta_{\min} = 2.4$ °  
 $h = -11 \rightarrow 11$   
 $k = -7 \rightarrow 7$   
 $l = -17 \rightarrow 13$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.035$   
 $wR(F^2) = 0.088$   
 $S = 0.83$   
3167 reflections  
164 parameters  
1 restraint  
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map  
Hydrogen site location: inferred from neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0455P)^2]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.021$   
 $\Delta\rho_{\max} = 0.39$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.38$  e Å<sup>-3</sup>  
Absolute structure: Flack (1983), 1405 Friedel pairs  
Flack parameter: 0.00 (8)

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|      | x            | y            | z            | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|--------------|--------------|----------------------------------|
| S1   | 0.19325 (6)  | 0.31172 (10) | 0.75706 (4)  | 0.02115 (14)                     |
| S2   | 0.22445 (7)  | 0.29779 (13) | 0.39671 (4)  | 0.03087 (17)                     |
| O1   | 0.15911 (18) | 0.3941 (3)   | 0.64203 (11) | 0.0213 (4)                       |
| O2   | 0.1803 (2)   | 0.0753 (3)   | 0.75046 (13) | 0.0330 (5)                       |
| O3   | 0.33583 (19) | 0.4105 (3)   | 0.80406 (12) | 0.0278 (4)                       |
| C1   | 0.3827 (2)   | 0.3018 (6)   | 0.33242 (16) | 0.0278 (5)                       |
| H1   | 0.4116       | 0.1822       | 0.2915       | 0.033*                           |
| C2   | 0.4623 (3)   | 0.4953 (5)   | 0.34661 (18) | 0.0275 (6)                       |
| H2   | 0.5542       | 0.5244       | 0.3163       | 0.033*                           |
| C3   | 0.3976 (2)   | 0.6547 (4)   | 0.41104 (16) | 0.0193 (5)                       |
| H3   | 0.4385       | 0.7991       | 0.4282       | 0.023*                           |
| C4   | 0.2621 (3)   | 0.5599 (4)   | 0.44481 (16) | 0.0201 (5)                       |
| C5   | 0.1556 (3)   | 0.6739 (4)   | 0.51055 (16) | 0.0263 (6)                       |
| H5A  | 0.1611       | 0.8368       | 0.4993       | 0.032*                           |
| H5B  | 0.0465       | 0.6263       | 0.4870       | 0.032*                           |
| C6   | 0.1919 (3)   | 0.6291 (4)   | 0.62353 (17) | 0.0221 (5)                       |
| H6A  | 0.3035       | 0.6624       | 0.6477       | 0.027*                           |
| H6B  | 0.1265       | 0.7255       | 0.6617       | 0.027*                           |
| C7   | 0.0366 (3)   | 0.4163 (4)   | 0.81555 (16) | 0.0202 (5)                       |
| C8   | 0.0527 (3)   | 0.6179 (4)   | 0.86654 (16) | 0.0207 (5)                       |
| H8   | 0.1479       | 0.6996       | 0.8707       | 0.025*                           |
| C9   | -0.0732 (3)  | 0.7003 (4)   | 0.91203 (16) | 0.0232 (5)                       |
| H9   | -0.0638      | 0.8400       | 0.9465       | 0.028*                           |
| C10  | -0.2126 (3)  | 0.5795 (5)   | 0.90745 (16) | 0.0269 (6)                       |
| C11  | -0.2247 (3)  | 0.3748 (4)   | 0.85539 (17) | 0.0261 (6)                       |
| H11  | -0.3190      | 0.2912       | 0.8518       | 0.031*                           |
| C12  | -0.1017 (2)  | 0.2926 (5)   | 0.80928 (15) | 0.0233 (5)                       |
| H12  | -0.1110      | 0.1540       | 0.7739       | 0.028*                           |
| C13  | -0.3486 (3)  | 0.6686 (5)   | 0.9550 (2)   | 0.0415 (8)                       |
| H13A | -0.4457      | 0.6455       | 0.9078       | 0.062*                           |
| H13B | -0.3336      | 0.8284       | 0.9687       | 0.062*                           |
| H13C | -0.3554      | 0.5901       | 1.0197       | 0.062*                           |

Atomic displacement parameters ( $\text{\AA}^2$ )

|    | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$    | $U^{13}$    | $U^{23}$     |
|----|-------------|-------------|-------------|-------------|-------------|--------------|
| S1 | 0.0254 (3)  | 0.0184 (3)  | 0.0206 (3)  | 0.0022 (3)  | 0.0066 (2)  | 0.0007 (3)   |
| S2 | 0.0249 (3)  | 0.0325 (4)  | 0.0362 (3)  | -0.0035 (4) | 0.0080 (2)  | -0.0007 (4)  |
| O1 | 0.0264 (9)  | 0.0206 (9)  | 0.0176 (8)  | -0.0047 (7) | 0.0059 (7)  | -0.0013 (6)  |
| O2 | 0.0464 (11) | 0.0142 (11) | 0.0429 (12) | 0.0044 (8)  | 0.0224 (9)  | 0.0026 (8)   |
| O3 | 0.0240 (9)  | 0.0355 (11) | 0.0228 (9)  | 0.0015 (8)  | -0.0004 (7) | 0.0032 (8)   |
| C1 | 0.0266 (12) | 0.0343 (14) | 0.0230 (11) | 0.0024 (15) | 0.0053 (9)  | -0.0027 (15) |
| C2 | 0.0210 (13) | 0.0330 (16) | 0.0301 (13) | 0.0011 (11) | 0.0097 (10) | 0.0086 (12)  |
| C3 | 0.0179 (11) | 0.0230 (15) | 0.0163 (11) | 0.0085 (10) | 0.0000 (8)  | 0.0010 (10)  |

## supplementary materials

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|     |             |             |             |              |             |              |
|-----|-------------|-------------|-------------|--------------|-------------|--------------|
| C4  | 0.0235 (12) | 0.0206 (14) | 0.0160 (11) | 0.0033 (10)  | 0.0018 (9)  | 0.0051 (10)  |
| C5  | 0.0282 (13) | 0.0274 (16) | 0.0245 (12) | 0.0073 (11)  | 0.0080 (10) | 0.0070 (11)  |
| C6  | 0.0246 (12) | 0.0189 (14) | 0.0236 (12) | -0.0014 (10) | 0.0056 (9)  | -0.0005 (10) |
| C7  | 0.0245 (13) | 0.0214 (14) | 0.0149 (11) | 0.0018 (10)  | 0.0032 (9)  | 0.0009 (10)  |
| C8  | 0.0277 (13) | 0.0196 (14) | 0.0149 (11) | -0.0032 (10) | 0.0041 (9)  | 0.0017 (9)   |
| C9  | 0.0317 (13) | 0.0192 (13) | 0.0185 (11) | 0.0031 (11)  | 0.0026 (9)  | -0.0029 (11) |
| C10 | 0.0283 (14) | 0.0343 (17) | 0.0182 (12) | 0.0025 (12)  | 0.0036 (10) | -0.0009 (11) |
| C11 | 0.0251 (13) | 0.0309 (17) | 0.0225 (12) | -0.0049 (10) | 0.0039 (10) | -0.0049 (10) |
| C12 | 0.0300 (12) | 0.0200 (13) | 0.0196 (10) | -0.0027 (13) | 0.0023 (8)  | -0.0010 (12) |
| C13 | 0.0290 (14) | 0.059 (2)   | 0.0373 (15) | 0.0042 (15)  | 0.0080 (11) | -0.0180 (16) |

### *Geometric parameters (Å, °)*

|          |             |             |             |
|----------|-------------|-------------|-------------|
| S1—O2    | 1.4236 (19) | C6—H6A      | 0.9900      |
| S1—O3    | 1.4250 (17) | C6—H6B      | 0.9900      |
| S1—O1    | 1.5776 (15) | C7—C8       | 1.379 (3)   |
| S1—C7    | 1.758 (2)   | C7—C12      | 1.396 (3)   |
| S2—C1    | 1.700 (2)   | C8—C9       | 1.399 (3)   |
| S2—C4    | 1.708 (3)   | C8—H8       | 0.9500      |
| O1—C6    | 1.464 (3)   | C9—C10      | 1.396 (3)   |
| C1—C2    | 1.347 (4)   | C9—H9       | 0.9500      |
| C1—H1    | 0.9500      | C10—C11     | 1.402 (4)   |
| C2—C3    | 1.438 (3)   | C10—C13     | 1.500 (3)   |
| C2—H2    | 0.9500      | C11—C12     | 1.382 (3)   |
| C3—C4    | 1.423 (3)   | C11—H11     | 0.9500      |
| C3—H3    | 0.9500      | C12—H12     | 0.9500      |
| C4—C5    | 1.507 (3)   | C13—H13A    | 0.9800      |
| C5—C6    | 1.497 (3)   | C13—H13B    | 0.9800      |
| C5—H5A   | 0.9900      | C13—H13C    | 0.9800      |
| C5—H5B   | 0.9900      |             |             |
| O2—S1—O3 | 119.74 (12) | C5—C6—H6A   | 110.0       |
| O2—S1—O1 | 104.53 (10) | O1—C6—H6B   | 110.0       |
| O3—S1—O1 | 108.65 (9)  | C5—C6—H6B   | 110.0       |
| O2—S1—C7 | 108.87 (12) | H6A—C6—H6B  | 108.4       |
| O3—S1—C7 | 109.28 (11) | C8—C7—C12   | 121.5 (2)   |
| O1—S1—C7 | 104.69 (10) | C8—C7—S1    | 119.52 (18) |
| C1—S2—C4 | 92.64 (13)  | C12—C7—S1   | 119.0 (2)   |
| C6—O1—S1 | 116.46 (14) | C7—C8—C9    | 119.0 (2)   |
| C2—C1—S2 | 111.8 (2)   | C7—C8—H8    | 120.5       |
| C2—C1—H1 | 124.1       | C9—C8—H8    | 120.5       |
| S2—C1—H1 | 124.1       | C10—C9—C8   | 120.8 (2)   |
| C1—C2—C3 | 115.0 (2)   | C10—C9—H9   | 119.6       |
| C1—C2—H2 | 122.5       | C8—C9—H9    | 119.6       |
| C3—C2—H2 | 122.5       | C9—C10—C11  | 118.7 (2)   |
| C4—C3—C2 | 108.6 (2)   | C9—C10—C13  | 120.9 (3)   |
| C4—C3—H3 | 125.7       | C11—C10—C13 | 120.3 (2)   |
| C2—C3—H3 | 125.7       | C12—C11—C10 | 121.1 (2)   |
| C3—C4—C5 | 126.0 (2)   | C12—C11—H11 | 119.5       |
| C3—C4—S2 | 111.98 (17) | C10—C11—H11 | 119.5       |

|             |              |                 |              |
|-------------|--------------|-----------------|--------------|
| C5—C4—S2    | 121.96 (18)  | C11—C12—C7      | 118.9 (3)    |
| C6—C5—C4    | 115.19 (19)  | C11—C12—H12     | 120.5        |
| C6—C5—H5A   | 108.5        | C7—C12—H12      | 120.5        |
| C4—C5—H5A   | 108.5        | C10—C13—H13A    | 109.5        |
| C6—C5—H5B   | 108.5        | C10—C13—H13B    | 109.5        |
| C4—C5—H5B   | 108.5        | H13A—C13—H13B   | 109.5        |
| H5A—C5—H5B  | 107.5        | C10—C13—H13C    | 109.5        |
| O1—C6—C5    | 108.57 (19)  | H13A—C13—H13C   | 109.5        |
| O1—C6—H6A   | 110.0        | H13B—C13—H13C   | 109.5        |
| O2—S1—O1—C6 | -169.41 (15) | O3—S1—C7—C8     | 22.3 (2)     |
| O3—S1—O1—C6 | -40.48 (18)  | O1—S1—C7—C8     | -93.96 (19)  |
| C7—S1—O1—C6 | 76.18 (17)   | O2—S1—C7—C12    | -25.2 (2)    |
| C4—S2—C1—C2 | 0.0 (2)      | O3—S1—C7—C12    | -157.67 (17) |
| S2—C1—C2—C3 | -0.4 (3)     | O1—S1—C7—C12    | 86.1 (2)     |
| C1—C2—C3—C4 | 0.7 (3)      | C12—C7—C8—C9    | -0.6 (3)     |
| C2—C3—C4—C5 | -177.6 (2)   | S1—C7—C8—C9     | 179.48 (17)  |
| C2—C3—C4—S2 | -0.6 (2)     | C7—C8—C9—C10    | 0.9 (3)      |
| C1—S2—C4—C3 | 0.37 (17)    | C8—C9—C10—C11   | -0.6 (3)     |
| C1—S2—C4—C5 | 177.50 (18)  | C8—C9—C10—C13   | -179.2 (2)   |
| C3—C4—C5—C6 | -94.9 (3)    | C9—C10—C11—C12  | 0.0 (3)      |
| S2—C4—C5—C6 | 88.4 (2)     | C13—C10—C11—C12 | 178.6 (2)    |
| S1—O1—C6—C5 | 178.83 (14)  | C10—C11—C12—C7  | 0.3 (3)      |
| C4—C5—C6—O1 | -67.3 (3)    | C8—C7—C12—C11   | 0.0 (3)      |
| O2—S1—C7—C8 | 154.71 (18)  | S1—C7—C12—C11   | 179.91 (17)  |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H... <i>A</i>      | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|------------------------------|-------------|---------------|-----------------------|-------------------------|
| C2—H2...O2 <sup>i</sup>      | 0.95        | 2.58          | 3.523 (3)             | 174                     |
| C6—H6B...O2 <sup>ii</sup>    | 0.99        | 2.41          | 3.161 (3)             | 132                     |
| C13—H13A...O3 <sup>iii</sup> | 0.98        | 2.58          | 3.496 (3)             | 155                     |

Symmetry codes: (i)  $-x+1, y+1/2, -z+1$ ; (ii)  $x, y+1, z$ ; (iii)  $x-1, y, z$ .

Fig. 1

